

THE STEREOSELECTIVE TOTAL SYNTHESIS OF  
RACEMIC  $\gamma$ -EUDESMOL

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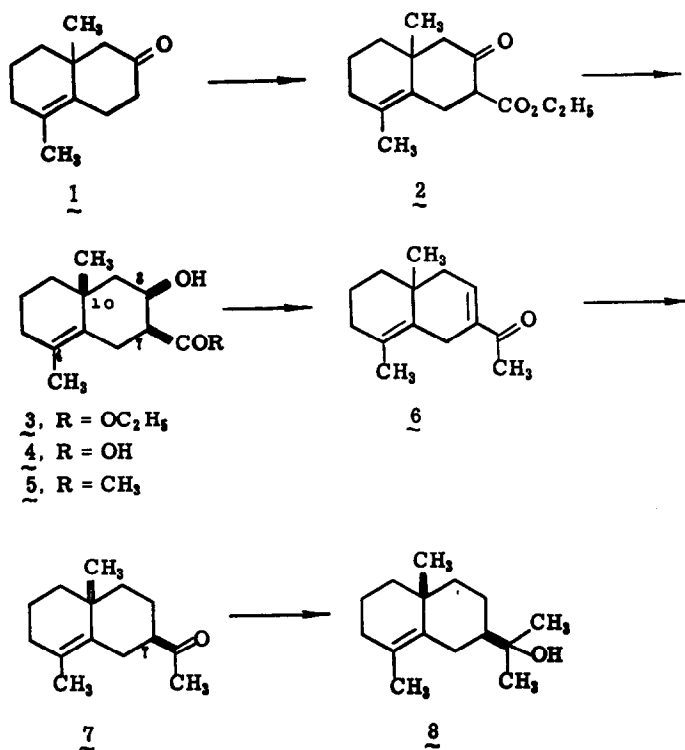
Certain projects currently underway in our laboratory called for a supply of pure  $\gamma$ -eudesmol.<sup>1</sup> A recent report by Pinder and Williams<sup>2</sup> suggests that this material as found in nature cannot be easily purified. These workers synthesized a pure sample of the naturally occurring (+)-form of  $\gamma$ -eudesmol starting from (+)- $\alpha$ -cyperone.<sup>3</sup> In view of the scarcity of this latter substance, we decided to investigate potential routes to the racemic form of  $\gamma$ -eudesmol (8).<sup>†</sup> From these investigations an efficient total synthesis of racemic  $\gamma$ -eudesmol (9) has evolved whose essential features are outlined below.

The known ketone 1<sup>4</sup> was converted by treatment with sodium hydride in diethyl carbonate to the corresponding  $\beta$ -keto ester 2 [bp 118° (bath temp.) at 0.05 mm]. Selective reduction ( $\text{KBH}_4$ )

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† All formulas represent racemic substances even though only one enantiomer is depicted.



afforded hydroxy ester 3 [ $\lambda_{\text{max}}^{\text{film}}$  2.85 (OH), 5.78 (CO), 8.35, 9.10, 9.60, and 10.50  $\mu$ ] which, in turn, yielded hydroxy acid 4 [mp 140-142°,  $\lambda_{\text{max}}^{\text{KBr}}$  2.90 (OH), 3.0-4.0 (COOH), 5.85 (CO), 8.20, 8.50, 8.70, 10.50, and 11.25  $\mu$ ;  $\delta_{\text{TMS}}^{\text{CDCl}_3}$  7.05 (OH; 2H), 4.43 (H-8,  $W_{1/2} = 7\text{Hz}$ ), 1.67 (C-4  $\text{CH}_3$ ), and 1.26 ppm (C-10  $\text{CH}_3$ ); 60% yield based on ketone 1] when saponified with ethanolic potassium hydroxide. The nmr spectrum of this acid supports the assigned C-8 stereochemistry;<sup>4</sup> the orientation depicted for C-7 follows from stability considerations.

The crude hydroxy ketone 5 [ $\lambda_{\max}^{\text{film}}$  2.94 (OH), 5.83 (CO), 8.48, and 9.51  $\mu$ ] which resulted from addition of ethereal methylolithium to hydroxy acid 4, smoothly dehydrated in refluxing aqueous ethanolic potassium hydroxide giving unsaturated ketone 6 [bp 95<sup>0</sup> (bath temp.) at 0.05 mm;  $\lambda_{\max}^{\text{film}}$  6.00 (CO), 8.00, 9.32, 10.47, 11.20, and 12.48  $\mu$ ;  $\delta_{\text{TMS}}^{\text{CCl}_4}$  6.78 (H-8), 2.20 (CH<sub>3</sub> CO), 1.62 (C-4 CH<sub>3</sub>), and 0.96 ppm (C-10 CH<sub>3</sub>);  $\lambda_{\max}^{\text{EtOH}}$  232 m $\mu$  ( $\epsilon$  10,000)]. Catalytic hydrogenation over 5% palladium-on-carbon in ethanolic potassium hydroxide<sup>5</sup> afforded the dihydro derivative 7 [bp 110<sup>0</sup> (bath temp.) at 0.05 mm;  $\delta_{\text{TMS}}^{\text{CCl}_4}$  2.06 (CH<sub>3</sub> CO), 1.61 (C-4 CH<sub>3</sub>), and 1.04 ppm (C-10 CH<sub>3</sub>)]. The gas chromatogram<sup>6</sup> of this material revealed about 8% of a second component, presumably the C-7 epimer, which could be completely removed by preparative gas chromatography. Equilibration in refluxing methanolic sodium carbonate afforded a 10:1 mixture of epimeric ketones, predominantly 7. Racemic  $\gamma$ -eudesmol (8) [bp 100<sup>0</sup> (bath temp.) at 0.04 mm;  $\lambda_{\max}^{\text{film}}$  2.95 (OH), 8.30, 8.66, 8.78, 10.72, 10.93, 11.80, 12.20, and 12.98  $\mu$ ;  $\delta_{\text{TMS}}^{\text{CCl}_4}$  1.58 (C-4 CH<sub>3</sub>), 1.13 (isopropyl CH<sub>3</sub>'s), and 1.01 ppm (C-10 CH<sub>3</sub>)<sup>1</sup>; 40% based on hydroxy acid 4] was secured via addition of ethereal methylolithium to ketone 7. The infrared spectrum and gas chromatographic retention time were identical with those of the natural (+)-isomer.

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## REFERENCES

1. R. B. Bates and E. K. Hendrickson, Chem. and Ind., 1759 (1962) and references therein.
2. A. R. Pinder and R. A. Williams, J. Chem. Soc., 2773 (1963).
3. R. Howe and F. J. McQuillin, ibid., 2423 (1955).
4. Cf. J. A. Marshall and N. Cohen, J. Am. Chem. Soc., 87, 2773 (1965); J. A. Marshall, N. Cohen, and A. R. Hochstetler, ibid., 88, 3408 (1966).
5. Cf. R. Howe and F. J. McQuillin, J. Chem. Soc., 1194 (1958).
6. A 10% DC-550 Silicone oil on 60-80 Chromosorb W column was used.